

## Role of nitrate in photosynthetic electron transport of *Chlorella vulgaris*

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### Abstract

Addition of nitrate to a suspension of  $\text{NO}_3^-$ -depleted *Chlorella vulgaris* cells raised the  $\text{O}_2$ -evolving capacity of the organism by 60 %. The rate of  $\text{O}_2$ -evolution under flash irradiation of the depleted cells was drastically reduced, which could be restored by addition of  $\text{NO}_3^-$ . The 2,5-dibromo-3-methyl-6-isopropyl-*p*-benzoquinone (DBMIB)-insensitive  $\text{O}_2$ -evolution, *i.e.*, photosystem (PS) 2 activity of  $\text{NO}_3^-$ -depleted cells, showed a 75 % stimulation by addition of  $\text{NO}_3^-$ . PS1-mediated electron transport was also stimulated (50 %) by addition of  $\text{NO}_3^-$ . Fluorescence yields of the  $\text{NO}_3^-$ -depleted cells were significantly reduced. A normal fluorescence response was restored by the addition of  $\text{NO}_3^-$ . The fluorescence yield of the  $\text{NO}_3^-$ -depleted and DCMU-treated-cells increased significantly after addition of  $\text{NO}_3^-$  ions, indicating a further reduction of the primary acceptor of PS2 (Q). In addition, the low temperature fluorescence emission spectra showed that energy transfer to PS2 and PS1 was much higher when nitrate was present. Hence nitrate accelerates the light-induced charge transfer from the intact  $\text{O}_2$ -evolving system to the primary electron acceptor of PS2 and stimulates the PS1-mediated electron transport.

*Additional key words:* chlorophyll fluorescence; DCMU; 2,5-dibromo-3-methyl-6-isopropyl-*p*-benzoquinone; oxygen evolution rate; photosystem 1 and 2 activities.

### Introduction

The enhanced  $\text{O}_2$ -evolution of irradiated *Chlorella* cells by addition of  $\text{NO}_3^-$  was first observed by Warburg and Negelein (1920). The increased  $\text{O}_2$ -evolution in response to  $\text{NO}_3^-$ -addition was attributed to an accelerated  $\text{NO}_3^-$ -reduction (Losada and Guerrero 1979). Using thylakoid preparations, which under the chosen

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*Abbreviations:* Chl - chlorophyll; DBMIB - 2,5-dibromo-3-methyl-6-isopropyl-*p*-benzoquinone; DCMU - 3-(3,4)-dichlorophenyl)-1,1-dimethylurea; DCPIP - 2,6-dichlorophenol indophenol; DCQ - dichloro-*p*-benzoquinone, MV - methylviologen;  $\text{NH}_2\text{OH}$  - hydroxylamine; PS - photosystem.

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experimental condition could not reduce nitrate any more, the involvement of  $\text{NO}_3^-$  ions at the donor side of PS2 was demonstrated (Osman *et al.* 1982): they reported that nitrate did not activate the PS1-activity. This conclusion disagrees with the interpretation of Warburg *et al.* (1965), who considered nitrate as possible electron acceptor for Hill reaction. In *Chlorella*, nitrogen starvation decreases yield of PS2 and photosynthetic efficiency (Sayed 1998). Marek and Beranová (1989) demonstrated that  $\text{NO}_3^-$  (up to 20 mM) increased the photosynthetic rate, ribulose-1,5-bisphosphate carboxylase/oxygenase activity, and transpiration rate of spring barley.  $\text{NO}_3^-$  assimilation uses photosynthetically derived reductant ( $\text{NADPH}_2$ ) and stimulates the rate of non-cyclic electron flow by acting as a second electron accepting process in addition to  $\text{CO}_2$ -fixation (De la Torre *et al.* 1991).

The present work was designed using partially  $\text{NO}_3^-$ -depleted *C. vulgaris* as intact system to reevaluate the  $\text{NO}_3^-$  effect on the photosynthetic electron transport.

## Materials and methods

*Chlorella vulgaris*, strain (211-11b) (Sammlung von Algenkulturen, Pflanzen-physiologisches Institut, Universität Göttingen, Germany) mass culture was grown as described by Lorenzen (1964). The culture grew under 12 fluorescent tubes (irradiance of  $120 \text{ W m}^{-2}$ ) at  $25 \pm 1^\circ \text{C}$  and continuous aeration with a mixture of 97 % air and 3 %  $\text{CO}_2$ . Chlorophyll (Chl) content was determined according to Metzner *et al.* (1965). Algal cells were harvested during the exponential phase (5 d). In order to obtain partially  $\text{NO}_3^-$ - and  $\text{Cl}^-$ -depleted cells, they were centrifuged at  $3\,000 \times g$ , washed several times with bidistilled water and then in 3 mM  $\text{NH}_4\text{SO}_4$  solution (to minimize the nitrate reductase activity), resuspended in phosphate buffer (pH 7.2), and finally irradiated by  $300 \text{ W m}^{-2}$  for one hour. The concentration of  $\text{NO}_3^-$  in the cells after this treatment, determined according to Allen *et al.* (1974), ranged between 0.2–0.3 mM. About 95 % of the oxygen evolution capacity of the  $\text{NO}_3^-$ -depleted cells could be restored by adding 10 mM  $\text{KNO}_3$ .

Oxygen evolution rate of a cell suspension (Chl content adjusted to  $15 \text{ g m}^{-3}$ ) under continuous irradiation ("white", 150 W tungsten lamps) was measured polarographically with a Clark-type electrode at  $20 \pm 1^\circ \text{C}$ . A dark adaptation time of 30 min after addition of  $\text{NO}_3^-$  was needed for 95 % restoration of the  $\text{O}_2$ -evolution capacity of the depleted cells. For measuring the PS1 activity (DCMU-insensitive  $\text{O}_2$  uptake), sonicated cells were used. For sonication, algal culture was centrifuged at  $3\,000 \times g$  for 3 min. The pellet was suspended in cooled nitrate-free medium, sonicated for 3 min at 15 amplitude micron with a *Sonniprep 150 MSE* equipped with 1.3 cm horn, and centrifuged again at  $5\,000 \times g$  for 5 min. The pellet was used for measuring PS1 activity (DCMU-insensitive  $\text{O}_2$  uptake) using sonicated algal cells equivalent to  $22 \text{ g(Chl) m}^{-3}$ . Flash induced oxygen yield was measured using a Joliot-type electrode (Joliot and Joliot 1968). According to this method,  $10 \text{ cm}^3$  algal suspension was centrifuged at  $3\,000 \times g$ , then resuspended in  $50 \text{ cm}^3$  phosphate buffer (pH 6.5). Sample aliquots of  $150 \text{ mm}^3$  each [ $50 \text{ g(Chl) m}^{-3}$ ] were placed on the electrode surface and irradiated by a xenon flash lamp with a sequence of 300 ms

flashes (4 Hz frequency) following 5 min dark adaptation with or without  $\text{NO}_3^-$ . The produced signals were evaluated on a storage oscilloscope (Hameg model MH 42) and printed by a Servogor recorder.

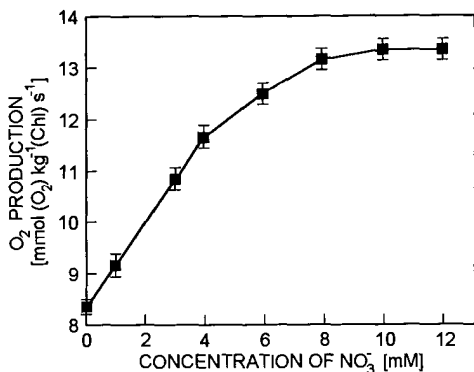


Fig. 1. Effect of different concentrations of  $\text{NO}_3^-$  ( $\text{KNO}_3$ ) on the  $\text{O}_2$  evolution of  $\text{NO}_3^-$ -depleted *Chlorella vulgaris* cells. Measurements in phosphate buffer (pH 6.8). The vertical bars represent the standard error of the mean values of at least five determinations.

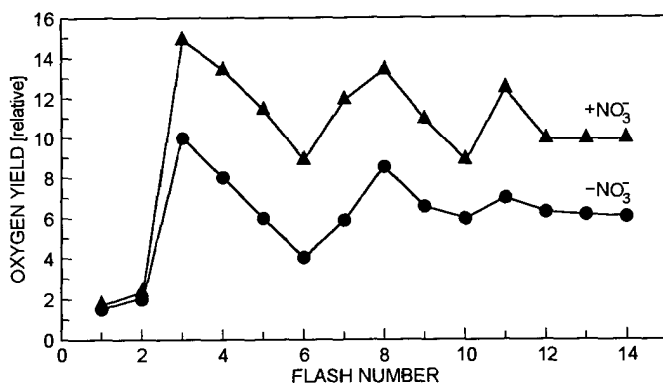


Fig. 2. Effect of  $\text{NO}_3^-$  (10 mM  $\text{KNO}_3$ ) on the flash oxygen yields of  $\text{NO}_3^-$ -depleted *Chlorella vulgaris* cells. The cells were suspended in 50 mM phosphate buffer, pH 6.5 and incubated for 3 min in the dark on the electrode surface prior to measurements.

Table 1. Effect of  $\text{NO}_3^-$  (10 mM  $\text{KNO}_3$ ) on PS2 (DBMIB-insensitive  $\text{O}_2$ -evolution in the presence of 1  $\mu\text{M}$  DBMIB and 1 mM DCQ) and PS1 activities ( $\text{O}_2$ -uptake from 40  $\mu\text{M}$  DCPIP to 2 mM MV in presence of 10  $\mu\text{M}$  DCMU and 2 mM sodium ascorbate) [ $\text{mmol(O}_2\text{) kg}^{-1} (\text{Chl}) \text{ s}^{-1}$ ] of nitrate-depleted *Chlorella vulgaris*. Means  $\pm$  standard errors (SE).

	- $\text{NO}_3^-$	+ $\text{NO}_3^-$	Stimulation [%]
PS2	4.4 $\pm$ 0.6	7.8 $\pm$ 0.6	75
PS1	2.2 $\pm$ 0.3	3.3 $\pm$ 0.6	50

Fluorescence transient measurements on algal suspension (Chl content  $10 \text{ g m}^{-3}$ ) were done at room temperature ( $22 \pm 1^\circ \text{C}$ ) according to Wiessner *et al.* (1981). The cells were centrifuged at  $3\,000\times g$ , resuspended in phosphate buffer (40 mM, pH 7.8), and then incubated with 10 mM  $\text{KNO}_3$  for 30 min before measurements. The blue actinic radiation was filtered by a 420 nm broad-band BG 38 Schott filter, switched on by a magnetic shutter (0.6 ms opening time), and focused on the sample cuvette. Fluorescence emission was filtered through two red cut-off filters (Rubilith-Amberlith ulano + Kodak Wratten no. 70) before reaching a RTCXP 1002 photomultiplier. Variations of the signal as a function of time were recorded on the memory screen of Tektronix DII 5103 N oscilloscope. A dark adaptation time of 10 min between measurements was sufficient for complete relaxation of the photosynthetic flow of algal cells between the light periods. Fluorescence emission spectra at liquid nitrogen temperatures were measured according to Harnischfeger (1977) using the cheese-cloth method (Cho *et al.* 1966).  $10 \text{ cm}^3$  of the  $\text{NO}_3^-$  and  $\text{Cl}^-$ -depleted algal suspension was centrifuged at  $3\,000\times g$ , then resuspended in 40 mM phosphate buffer (pH 7.2) to a standard Chl concentration ( $150 \text{ g m}^{-3}$ ). The samples were spotted carefully on two layers cheese cloth, fixed between two hard gummi rings, then chilled rapidly in liquid nitrogen ( $-196^\circ \text{C}$ ). The excitation radiation was a blue broad band at 470 nm. The fluorescent radiation was passed through photomultiplier, then printed by the Servogor recorder.

## Results and discussion

The addition of  $\text{NO}_3^-$  to  $\text{NO}_3^-$ - and  $\text{Cl}^-$ -depleted *C. vulgaris* cells stimulated the  $\text{O}_2$ -evolution capacity of the organism; this stimulation was concentration-dependent (Fig. 1). The maximum stimulation (60 %) was observed after addition of 10 mM  $\text{NO}_3^-$ .  $\text{O}_2$  emission from dark-adapted algae under flashes was also drastically raised by adding  $\text{NO}_3^-$  (Fig. 2). However, although  $\text{NO}_3^-$ -depleted cells evolved less  $\text{O}_2$  per flash than the nitrate-treated cells, both showed a similar pattern of  $\text{O}_2$  emission. Thus  $\text{NO}_3^-$  may interfere with an event associated with PS2, since the reduction in  $\text{O}_2$ -evolution under flashing irradiation indicates that electron transport to the pool of PQ is reduced in the absence of  $\text{NO}_3^-$ . This conclusion was supported by our results on the DBMIB-insensitive  $\text{O}_2$ -evolution (Table 1), which showed marked stimulation of  $\text{O}_2$ -evolution (75 %) in response to  $\text{NO}_3^-$  addition. PS1-mediated electron transport was also stimulated (by 50 %) after adding  $\text{NO}_3^-$  to  $\text{NO}_3^-$ -depleted cells (Table 1).

In order to localize the action sites of  $\text{NO}_3^-$  in photosynthetic electron transport, we studied the fluorescence induction kinetics at room temperature as well as the fluorescence emission spectra at liquid nitrogen temperature. The addition of  $\text{NO}_3^-$  to dark-adapted nitrate-depleted cells raised the fluorescence yield which resulted in a significant increase of the  $F_{\text{max}}$  value of variable fluorescence (Fig. 3A). Addition of DCMU to  $\text{NO}_3^-$ -depleted cells raised the fluorescence yield significantly, which was further increased by addition of  $\text{NO}_3^-$  (Fig. 3B). Because the yield of variable fluorescence is taken as an indicator of the redox state of Q (primary acceptor of

PS2) (Duysens and Sweers 1963, Butler 1966), the increased  $F_{\max}$  value means that  $\text{NO}_3^-$  leads to a stronger reduction of Q.

There are four possibilities for the accumulation of  $\text{Q}^-$ :

- (1) Blocking of the electron transport between the two photosystems by a poison such as DCMU.
- (2) Enhanced photoreduction of Q by increasing irradiance.
- (3) Chemical reduction by the addition of a reducing agent, e.g., sodium dithionate.
- (4) Stimulation of the electron transport at the donor side of PS2 to a level which brings the intermediate pool (A) between the two photosystems into a nearly completely reduced state.

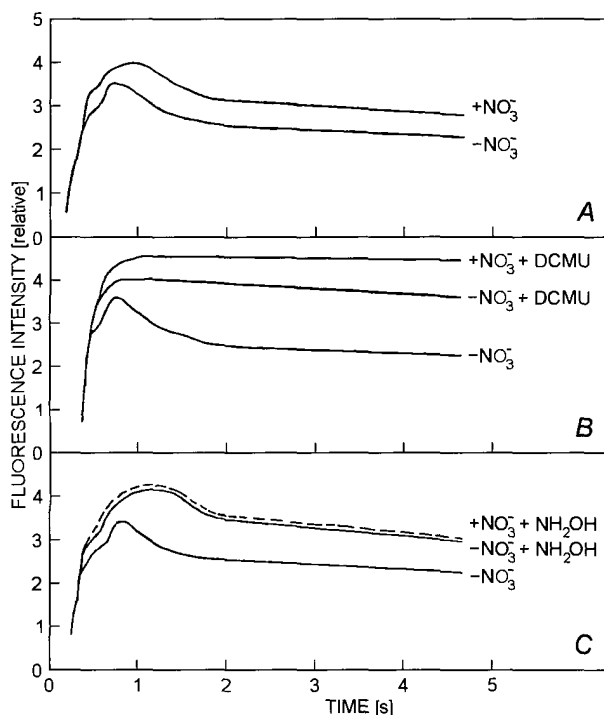


Fig. 3. Fluorescence induction kinetics of  $\text{NO}_3^-$ -depleted *Chlorella vulgaris* cells: (A) Before and after addition of 10 mM  $\text{KNO}_3$ . (B) Nitrate-depleted cells,  $\text{NO}_3^-$ -depleted + 10  $\mu\text{M}$  DCMU, and  $\text{NO}_3^-$ -depleted + 10  $\mu\text{M}$  DCMU + 10 mM  $\text{KNO}_3$ . (C)  $\text{NO}_3^-$ -depleted cells,  $\text{NO}_3^-$ -depleted cells + 25 mM  $\text{NH}_2\text{OH}$  added before the dark adaptation period, and  $\text{NO}_3^-$ -depleted cells + 25 mM  $\text{NH}_2\text{OH}$  + 10 mM  $\text{KNO}_3$ . Experimental conditions as described in Materials and methods.

Our results eliminated the first two possibilities. For the third possibility there was no evidence which could indicate a direct reduction of Q by nitrate. The relative position of the redox potentials of  $\text{Q}/\text{Q}^-$ , respectively  $\text{NO}_3^-/\text{NO}_3$ , makes this assumption unlikely. Hence we have to accept the fourth possibility, which is supported by our results on the fluorescence yield of DCMU-treated cells. The increase in  $F_{\max}$  of these cells after  $\text{NO}_3^-$  addition indicates that this anion stimulates

the charge transfer from the intact  $O_2$ -evolving system to the primary acceptor of PS2. In other words, nitrate stimulates the electron flow from the oxidizing side of PS2 to the primary electron acceptor Q, causing increase in the fluorescence yield.

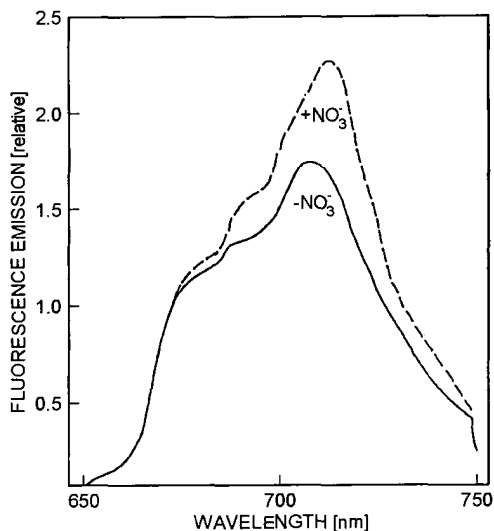


Fig. 4. Fluorescence emission spectra of  $NO_3^-$ -depleted *Chlorella vulgaris* cells (—) and of these cells after addition of 10 mM  $KNO_3$  (---) at liquid nitrogen temperature. Excited by broad blue band at 470 nm. Experimental conditions as described in Materials and methods.

To localize the site of  $NO_3^-$  action on the donor side of PS2 we used hydroxylamine as artificial electron donor for PS2, which donates its electron to the donor side of PS2 thus inducing strong reduction of Q. As shown in Fig. 3C, the induced fluorescence caused by addition of hydroxylamine was not affected by addition of  $NO_3^-$  and thus the role of  $NO_3^-$  is effective only with a natural electron donor of PS2 ( $X-H_2O$ ), i.e., the site of  $NO_3^-$  action is before the site of electron donation by hydroxylamine at the donor side of PS2.

The influence of  $NO_3^-$  on the efficiency of distribution of excitation energy between the two photosystems could be seen from the fluorescence emission spectra at liquid nitrogen temperature (Fig. 4). At this temperature,  $F_{718}$  is assigned to PS1 and  $F_{693}$  to PS2. The  $NO_3^-$  addition to  $NO_3^-$ -depleted cells raised the fluorescence yield of both photosystems indicating that the energy transfer to PS2 and PS1 is much higher when nitrate is present and showing an involvement of this anion in the energy transfer mechanism of the two photosystems.

In conclusion, our results suggest that in addition to the unique role of  $NO_3^-$  in nitrogen metabolism of higher plants and algae it acts as a stimulating anion similar to  $Cl^-$  (Kelley and Izawa 1978) and  $HCO_3^-$  (Stemler 1982, Mende and Wiessner 1985) on the photosynthetic electron transport at the two photosystems. The action site of this anion in PS2 is somehow engaged in the electron release on the PS2-donor side before the action site of hydroxylamine as electron donor.

## References

- Allen, F.E., Grimshaw, H.M., Parainson, J.A., Quarmby, C.: Chemical Analysis of Ecological Material. - Blackwell, London 1974.
- Butler, W.L.: Fluorescence yield in photosynthetic systems and its relation to electron transport. - In: Sanadi, D.R. (ed.): Current Topics in Bioenergetics. Vol. 1. Pp. 49-73. Acad. Press, New York 1966.
- Cho, F., Spencer, J., Govindjee: Emission spectra of *Chlorella* at very low temperature (-296° to -196°). - Biochim. biophys. Acta 126: 174-176, 1966.
- De la Torre, A., Delgado, B., Lara, C.: Nitrate dependent O<sub>2</sub> evolution in intact leaves. - Plant Physiol. 96: 898-901, 1991.
- Duysens, L.N.M., Sweers, H.E.: Mechanism of two photochemical reactions in algae by means of fluorescence. - In: Studies on Microalgae and Photosynthetic Bacteria. Pp. 353-372. University of Tokyo Press, Tokyo 1963.
- Harnischfeger, G.: The use of fluorescence emission at 77 °K in the analysis of the photosynthetic apparatus of higher plants and algae. - Adv. bot. Res. 5: 1-52, 1977.
- Joliot, P., Joliot, A.: A polarographic method for detection of oxygen production and reduction of Hill reagent by isolated chloroplasts. - Biochim. biophys. Acta 153: 625-634, 1968.
- Kelley, P.M., Izawa, S.: The role of chloride ion in photosystem II. I. Effects of chloride ion on photosystem II electron transport and on hydroxylamine inhibition. - Biochim. biophys. Acta 502: 198-210, 1978.
- Lorenzen, H.: Synchronization of *Chlorella* with light-dark changes and periodical dilution to a standard cell number. - In: Zeiten, E. (ed.): Synchrony of Cell Division and Growth. Pp. 571. Interscience, New York 1964.
- Losada, M., Guerrero, M.G.: The photosynthetic reduction of nitrate and its regulation. - In: Barber, J. (ed.): Photosynthesis in Relation to Model Systems. Pp. 365-408. Elsevier, Amsterdam - New York - Oxford 1979.
- Marek, M., Beranová, J.: [Effect of increasing concentrations of nitrates in nutrient medium on the photosynthetic characteristics of primary leaves of spring barley (*Hordeum vulgare* L.).] - Rostlinná Výroba (Praha) 35: 85-93, 1989. [In Czech.]
- Mende, D., Wiessner, W.: Bicarbonate *in vivo* requirement of photosystem II in the green alga *Chlamydomonas stellata*. - J. Plant Physiol. 118: 259-266, 1985.
- Metzner, H., Rau, H., Senger, H.: Untersuchungen zur Synchronisierbarkeit einzelner Pigmentmangelmutanten von *Chlorella*. - Planta 65: 186-194, 1965.
- Osman, M.E.H., Metzner, H., Fisher, K.: Effects of nitrate on thylakoid reactions I. Influence on photosynthetic electron transport. - Photosynthetica 16: 7-12, 1982.
- Sayed, O.H.: Analysis of photosynthetic responses and adaptation to nitrogen starvation in *Chlorella* using *in vivo* chlorophyll fluorescence. - Photosynthetica 35: 611-619, 1998.
- Stemler, A.: The functional role of bicarbonate in photosynthetic light reaction II. - In: Govindjee (ed.): Photosynthesis. Vol. II. Pp. 513-539. Acad. Press, New York - London - Paris - San Diego - San Francisco - São Paulo - Sydney - Tokyo - Toronto 1982.
- Warburg, O., Krippahl, G., Jetschmann, C.: Widerlegung der Photolyse des Wassers und Beweis der Photolyse der Kohlensäure nach Versuchen mit lebender *Chlorella* und den Hill-Reagentien Nitrat und K<sub>3</sub>Fe(CN)<sub>6</sub>. - Z. Naturforsch. 20b: 993-996, 1965.
- Warburg, O., Negelein, E.: Über die Reduktion der Salpetersäure in grünen Zellen. - Biochem. Z. 110: 66-115, 1920.
- Wiessner, W., Dubertret, G., Henry-Hiss, Y., Mende, D., Lefort-Tran, M.: Fluorescence transients as indicator for the existence of regulatory mechanisms in photosynthetic electron transport. - Ber. deutsch. bot. Ges. 94: 503-515, 1981.